

This Page Is Inserted by IFW Operations
and is not a part of the Official Record

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

**As rescanning documents *will not* correct images,
please do not report the images to the
Image Problem Mailbox.**

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification: A23L 1/304, A61K 7/16	A1	(11) International Publication Number: WO 99/08550 (43) International Publication Date: 25 February 1999 (25.02.99)
(21) International Application Number: PCT/EP98/05119 (22) International Filing Date: 11 August 1998 (11.08.98) (30) Priority Data: 9717598.8 19 August 1997 (19.08.97) GB (71) Applicant (for all designated States except US): SMITHKLINE BEECHAM PLC [GB/GB]; New Horizons Court, Brentford, Middlesex TW8 9EP (GB). (72) Inventor; and (75) Inventor/Applicant (for US only): PARKER, David, Myatt [GB/GB]; SmithKline Beecham Consumer Brands, The Royal Forest Factory, Coleford, Gloucestershire GL16 8JB (GB). (74) Agent: WHITE, Susan, Mary; SmithKline Beecham plc, Corporate Intellectual Property, Two New Horizons Court, Brentford, Middlesex TW8 9EP (GB).		(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, HR, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: SOLID COMPOSITION FOR REDUCING TOOTH EROSION (57) Abstract Solid or semi-solid acidic oral compositions having reduced tooth erosion characteristics are prepared by adding a calcium compound to an acid composition so that the mol ratio of calcium to acid ranges from 0.3 to 0.8, and the effective pH of the composition, if necessary after adjustment with an alkali, is from 3.5 to 4.5.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

SOLID COMPOSITION FOR REDUCING TOOTH EROSION

The present invention relates to compositions for oral use, in particular solid or
5 semi-solid acidic products and oral healthcare compositions, and to the use of
calcium in such compositions to alleviate or prevent the tooth damage associated
with the consumption of acid. In particular, the present invention alleviates
palatability problems associated with calcium addition to products.

10 It is thought that erosion of teeth is caused *inter alia* by acidic foodstuffs leaching
out calcium from the teeth faster than it can be replaced by normal remineralisation
processes. When a product is prepared in accordance with this invention, and
introduced into the oral cavity for consumption or healthcare purposes, the
dissolution or removal of calcium and phosphate from teeth by chemical processes is
15 significantly reduced.

Calcium is the most abundant mineral in the body. The vast majority of calcium is
deposited in the bones and teeth but the mineral is also essential for other bodily
functions such as the regulation of nerve function, the contraction of muscles and
20 clotting of blood. Calcium is a common constituent of beverages being derived
from fruit ingredients and from hard water when this is used in beverage production
without prior softening. Values for the concentration of calcium occurring in this
way are typically in the range 0.005-0.02 % w/w. Interest in the general nutritional
benefits of diet fortification by calcium ion has led to a search for practical ways to
25 incorporate this ion at higher levels from 0.02 % w/w to 2 % w/w. The use of
calcium as a supplement for beverages has been described in WO88/03762.

It is well known that the addition of malic acid will help maintain the solubility of
calcium in calcium fortified beverages therefore minimizing losses due to
30 precipitation. This is because of the formation of a soluble complex "calcium citrate
malate". On the other hand, Lussi et al (1995, Caries Res 29, 349-354) have

associated the titratable acidity of a beverage with its erosive potential; the greater the concentration of acid in the beverage the more damaging to teeth it became.

5 In EP 551398 (Procter & Gamble) there is disclosed a method for preventing the erosion of tooth enamel by consuming an acid beverage (having a pH of less than 5.5) comprising from 0.02% to 0.15% of calcium in the form of a calcium citrate malate complex having a molar ratio of citrate to malate of 1:0.5 to 1:4.5. In the calcium citrate malate complexes the molar ratio of total moles calcium:total moles citrate:total moles malate may be from about 2:1:1 to about 6:3:4. A preferred
10 complex for beverages has the molar ratio 4:2:3. US Patent 5,073,389 describes the use of calcium citrate malate to provide a mineral supplemented candy product.

We have found that inclusion of high levels of calcium in products gives palatability problems. However, we have found that effective reduction of tooth erosion in
15 acidic oral compositions can be achieved without impairing palatability using lower amounts of calcium relative to the acidulant when the pH of the composition is also controlled. WO 97/30601 (published 28 August 1997) discloses controlled pH liquid compositions containing calcium and an acidulant in a defined ratio.

20 The present invention provides a solid or semi solid composition for oral use containing a calcium compound and an acidulant characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.

25 The term effective pH is used in the context of the present invention to mean the pH of the composition before solidification (where the composition is prepared via a liquid phase intermediate) or the pH of the composition when reconstituted or dissolved in a liquid, eg. water. The term solidification encompasses the treatment
30 or supplementation of liquid phase intermediates to form a solid or semi-solid.

In another aspect, the present invention provides the use of calcium as a tooth erosion inhibitor in a solid or semi-solid acidic composition for oral administration comprising a calcium compound and an acidulant, characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.

In a further aspect, the present invention provides a method of reducing the tooth erosion potential of a solid or semi-solid acidic oral composition comprising adding calcium to the acidic oral composition so that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and the effective pH is from 3.5 to 4.5, obtaining an effective pH within the range 3.5 to 4.5 by addition of an alkali, if necessary or desired.

The invention also extends to a method of reducing tooth erosion caused by acid in orally administered compositions comprising orally administering a solid or semi-solid composition comprising a calcium compound and an acidulant, characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.

The invention further extends to the use of a solid or semi-solid composition comprising a calcium compound and an acidulant in the manufacture of a medicament for the reduction of tooth erosion caused by acid in orally administered compositions, characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.

In a still further aspect, the present invention provides a process for preparing a composition of this invention which comprises mixing a calcium compound with an acidulant so that calcium is present in the range of 0.3 to 0.8 mol per mol of acid

and the effective pH of the composition is from 3.5 to 4.5. If necessary or desired an effective pH within the range 3.5 to 4.5 can be obtained by addition of an alkali.

The present invention is particularly applicable to solid or semi solid acidic substances for oral consumption such as boiled sweets, candies, tablets, lozenges, lollies, chews, jellies, gums, drops, dry powder blends such as powdered drinks intended for dissolution, eg. in water, and the like. Semi solid products also include dairy products such as yoghurts and set or frozen drinks.

Suitably the composition before solidification is prepared and tested using the techniques described in WO 97/30601, the entire contents of which are herein incorporated by reference.

The effective pH for compositions of the invention is higher than normally associated with acid-based products for human consumption which typically have an effective pH of about pH 3 in order to maintain palatability associated with sharpness in taste. Practice of the present invention does not cause taste defects in products. Although an increase in effective pH to around pH 4 would be expected to reduce the sharpness in taste provided by the acidulant, surprisingly the inclusion of calcium in accordance with this invention mitigates this.

A further advantage arises from the use of low levels of calcium in accordance with this invention, suitably in the form of an alkaline salt. The buffering capacity of the formulation is reduced by partial neutralisation of the acid, which allows saliva to neutralise remaining acid residues in the mouth more rapidly.

The absolute concentration of calcium used in the compositions of the present invention is not critical as this will vary according to the nature and concentration of the acids present. The acid composition may contain organic and/or inorganic acids and may be supplemented with vitamins such as ascorbic acid. The calcium concentration may vary from 0.001 mol. per litre to more than 0.25 mol. per litre,

typically from 0.002 mol. per litre to 0.1 mol. per litre, suitably from 0.01 mol. per litre to 0.05 mol. per litre.

5 The calcium may be added in any suitable form, conveniently as a soluble salt such as calcium carbonate, calcium hydroxide, calcium citrate, calcium malate, calcium lactate, calcium chloride, calcium glycerophosphate or calcium formate or any other salt which minimises any adverse flavour contribution to the composition.

10 Compositions of the invention may be prepared by mixing the acid (e.g. citric acid) with its corresponding calcium salt (e.g. calcium citrate) or another calcium salt. It may be advantageous to mix the acid with an alkaline calcium salt such as calcium carbonate or calcium hydroxide thereby minimizing the concentration of acid applied to the formulation. The acid can also be mixed with inorganic calcium salts such as calcium chloride. The molar ratio of calcium to acid may be 0.3 to 0.75,
15 typically 0.3 to 0.7, more typically 0.3 to 0.65, suitably 0.3 to 0.60 and preferably 0.3 - 0.55 or 0.4 to 0.55. Most preferably the molar ratio is at least 0.4, and a value of about 0.5 has been found to be especially effective.

The effective pH of the formulation may be adjusted to the desired range by the
20 addition of the calcium compound to the appropriate proportion relative to the acid. If necessary, depending on the acid present, the effective pH may be further adjusted by the application of an alkali e.g. sodium hydroxide or a suitable salt for example sodium citrate, sodium malate or sodium lactate. The effective pH of the composition is preferably not more than 4, most preferably from 3.7 to 3.9.
25 Compositions with an effective pH of about 3.8 have been found to be especially effective.

Typically the acid concentration in compositions of the invention, for example the citric acid or malic acid concentration in a fruit-based product would be in the range
30 0.01 % w/w to 4 % w/w, suitably in the range 0.1 % w/w to 1 % w/w. Other potable

acids conventional for products of the invention may also be used, such as lactic acid. Mixtures of potable acids may be used.

5 In a preferred embodiment, the acid composition is based on a concentrate prepared from a natural fruit juice, such as blackcurrant juice, for example a flavoured syrup concentrate. The calcium may be added in a suitable form to the concentrate and the resulting composition is formed into a solid or semi-solid. Preferably the product contains reduced levels of sugar or carbohydrate or is of low calorie type containing intense sweeteners.

10

The oral composition may contain magnesium or other ions as adjuncts for remineralisation. It may also contain an effective amount of malic acid or potable salts thereof to maintain the solubility of the calcium so as to prevent or minimize the precipitation of insoluble calcium salts. Added malic acid may provide as little
15 as 10% of the total acidity of the beverage, the remainder of the acidity being provided by other, preferably naturally present, acids such as citric acid, or by ascorbic acid.

The invention may be applied in a variety of products based on concentrates, in
20 particular to health products containing blackcurrant juice or extract or added vitamins. The compositions are typically solidified according to known methods such as freezing, cooking, gelling or by the formation of solid or semi solid emulsions or gels. Suitable formulation techniques can be found in standard confectionary texts such as 'Sugar Confectionary Manufacture' by E. B. Jackson
25 (2nd Edition).

The invention is advantageously applied to products containing natural or added citric acid. The products may be unsweetened or sweetened with sugar or intense sweeteners such as saccharine, aspartyl phenyl alanyl methyl ester, or other
30 sweeteners known in the art. The products may also contain other conventional

additives such as sodium benzoate, sorbic acid, sodium metabisulfite, ascorbic acid, flavourings and colourings.

The products may be prepared by mixing the ingredients according to conventional methods. Ingredients may be dissolved in water or in hot water, if required, prior to addition to the other components. Typically concentrates are pasteurised.

The invention is illustrated by the following Examples:

10 Example 1

A concentrated product is initially prepared by mixing the ingredients as follows.

The calcium carbonate is added to the other ingredients as a final addition.

	Blackcurrant juice concentrate	SG 1.27	84 litre
	Aspartyl phenyl alanyl methyl ester *		1.15 Kg
15	Acesulfame K		1.8 Kg
	Ascorbic acid		0.8 Kg
	Sodium benzoate		0.325 Kg
	Sodium metabisulfite		0.145 Kg
	Blackcurrant flavouring		0.3 litre
20	Water	up to final volume	1000 litre
	Calcium carbonate		4.2 Kg

*sold as Aspartame (RTM)

The mol ratio of calcium : acid is 0.5

25

The concentrate is adjusted to pH 3.7 with sodium hydroxide solution.

In-vitro planometry tests can be performed on the concentrate formulations as follows. Flat dental enamel sections are exposed to test solutions having a pH of 3.85 (x5 dilution of concentrate with water) at a temperature of 37°C for 30 minutes. Erosive potential is evaluated by physical measurement of the depth of enamel lost during the procedure. Whereas a control formulation comprising 14 mM citric acid, pH 3.2 results in a loss of 4 microns of enamel and a further control formulation of 14 mM citric acid, pH 3.85, removes 1.8 microns, a test formulation with adjusted pH and added calcium comprising 14 mM citric acid, 7 mM calcium,

pH 3.85 removes only 0.17 microns of enamel, demonstrating the utility of the invention.

This solution or the concentrate before dilution can be solidified according to Example 5 or 6 below.

5

Example 2

A solution was prepared by mixing ingredients as follows:

Ingredients		%w/v
Sodium benzoate		0.01
10 Malic acid		0.30
Flavouring		0.1
Artificial sweetener		0.05
Water	by difference	99.5
15 Calcium hydroxide		0.083

The resultant pH of the composition is typically 3.85 and has a calcium to acid molar ratio of 0.5. This solution can be solidified according to Example 5 or 6 below.

In vitro planometry tests were performed on the solution in which flat dental enamel

20 sections were exposed to test solutions at a temperature of 37°C for 30 minutes.

Erosive potential was evaluated by physical measurement of the depth of enamel lost during the procedure. Whereas a control formulation lacking the addition of calcium hydroxide gave a pH of 2.5 and resulted in a loss of 8.1 microns of enamel and a further control formulation in which the pH had been increased to pH 3.85
25 with sodium hydroxide removed 1.65 microns, the composition detailed above removed only 0.6 microns of enamel, demonstrating its utility in reducing tooth erosion.

Example 3

30 A solution was prepared by mixing ingredients as follows:

Ingredients		%w/w
Sugar		10
Sodium benzoate		0.01
Orange juice		5.04
35 Ascorbic acid		0.03
Citric acid monohydrate		0.15

	Flavouring	0.005
	Colouring	0.004
	Water	by difference
	Calcium carbonate	0.048
5	Sodium hydroxide	sufficient to adjust to pH 3.9
	Carbon dioxide	0.48

In this solution the mol ratio of calcium : acid is 0.46 (orange juice is typically 1 % w/w citric acid).

- 10 This solution is then solidified as further indicated in Example 5 or 6 below.

Example 4

A solution was prepared by mixing ingredients as follows:

	Ingredients	% w/w
15	Sugar	8
	Sodium benzoate	0.01
	Apple juice	10
	Ascorbic acid	0.03
	Malic acid	0.15
20	Flavouring	0.005
	Colouring	0.004
	Water	by difference
	Calcium carbonate	0.093
25	Sodium hydroxide	sufficient to adjust to pH 3.9.

In this solution the mol ratio of calcium : acid is 0.74 (apple juice is typically 0.6 % w/w malic acid). The solution is then solidified as indicated in the Example 5 or 6 below.

30 Example 5

Solidification techniques

Concentrates can be solidified by freezing e.g. at temperatures less than minus 5 degrees C, preferably at temperatures around minus 20 degrees C. Solutions can be boiled, e.g. for 10 minutes until a set point is reached, followed by cooling and

- 35 moulding if desired. Dissolved powder gelatine can be added (according to manufacturer's instructions) and the product allowed to set.

Example 6

Blackcurrant jellies

	Ingredients	grammes
5	Glucose syrup	564
	Gelatin 190 bloom	93
	Water	152
	Concentrate (e.g. Example 1)	191

10 Method

The glucose syrup is cooked to 85 % solids and the gelatine soaked in warm water to dissolve. The gelatine solution and concentrate is added to the glucose syrup solution. The mixture is moulded in cornflour and left overnight.

15 Example 7

Dry Powdered Orange Sports Drink

The ingredients are dry blended typically using a ribbon blender until a homogeneous mixture is obtained. The product is then filled into appropriate packaging, such as sachets, jars or drums.

20	Ingredients	kg
	Dextrose Monohydrate	389.12
	Maltodextrin	523.37
	Aspartame	0.58
	Acesulfame k	0.37
25	Tri-sodium citrate	16.54
	Sodium chloride	9.34
	Citric acid	36.97
	Ascorbic acid	1.17
	Potassium citrate	2.33
30	Calcium carbonate	11.46
	Orange flavour	2.92
	Beta-Carotene (1 %)	5.84
	Total	1000.00 kg

- 35 50g of the powder was dissolved in 500 ml of water to make an orange sports drink. The drink had a pH of 4 and a calcium to acid molar ratio of 0.6.

Example 8

Dry Powdered Low-Calorie Orange Sports Drink

The ingredients are dry blended typically using a ribbon blender until a homogeneous mixture is obtained. The product is then filled into appropriate packaging such as sachets, jars or drums.

	Ingredients	kg
	Maltodextrin	129.52
	Aspartame	30.73
	Acesulfame k	9.77
10	Tri-sodium citrate	153.07
	Sodium chloride	59.81
	Citric acid	353.23
	Ascorbic acid	27.55
	Potassium citrate	21.55
15	Calcium carbonate	109.50
	Orange flavour	35.09
	Beta-Carotene (1 %)	70.18
	Total	1000.00 kg

- 20 4g of the powder was dissolved in 500 ml of water to make a low-calorie orange sports drink. The drink had a pH of 4 and a calcium to acid molar ratio of 0.6.

25

CLAIMS

- 5 1. A solid or semi-solid composition for oral use containing a calcium compound and an acidulant characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the proportion of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.
- 10 2. A composition as claimed in claim 1 in which the calcium is present in the range 0.3 - 0.75 mol per mol of acid .
- 15 3. A composition as claimed in claim 1 in which the calcium is present in the range 0.3 - 0.65 mol per mol of acid .
- 20 4. A composition as claimed in claim 1 in which the calcium is present in the range 0.3 - 0.60 mol per mol of acid .
- 25 5. A composition as claimed in claim 1 in which the calcium is present in the range 0.3 - 0.55 mol per mol of acid .
6. A composition as claimed in any one of claims 1 to 5 in which the calcium is present in an amount of at least 0.4 mol per mol of acid .
7. A composition as claimed in any one of claims 1 to 6 in which the effective pH of the composition is not more than 4.
8. A composition as claimed in any one of claims 1 to 6 in which the effective pH is from 3.7 to 3.9.
- 30

9. A composition as claimed in any one of claims 1 to 8 in which the acid is citric acid, malic acid or lactic acid or mixtures thereof.
10. A composition as claimed in any one of claims 1 to 9 in which the calcium compound is calcium carbonate, calcium hydroxide, calcium citrate, calcium malate, calcium lactate, calcium chloride, calcium glycerophosphate or calcium formate.
11. A composition as claimed in any one of claims 1 to 10 which is a sweet.
12. A composition as claimed in claim 11 in which the sweet is a pastille.
13. A composition as claimed in any one of claims 1 to 10 which is a dry powder blend.
14. A composition as claimed in claim 13 which is a powdered drink product.
15. A composition as claimed in any one of claims 1 to 14 which is an oral healthcare composition.
16. Use of calcium as a tooth erosion inhibitor in a solid or semi-solid composition for oral administration comprising a calcium compound and an acidulant, characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.
17. Use as claimed in claim 16 in which the calcium is present in the range 0.3 - 0.75 mol per mol of acid .
18. Use as claimed in claim 16 in which the calcium is present in the range 0.3 - 0.65 mol per mol of acid .

19. Use as claimed in claim 16 in which the calcium is present in the range 0.3 - 0.60 mol per mol of acid.

5 20. Use as claimed in claim 16 in which the calcium is present in the range 0.3 - 0.55 mol per mol of acid.

21. Use as claimed in any one of claims 16 to 20 in which the calcium is present in an amount of at least 0.4 mol per mol of acid.

10

22. Use as claimed in any one of claims 16 to 21 in which the effective pH of the composition is not more than 4.

15 23. Use as claimed in any one of claims 16 to 22 in which the effective pH is from 3.7 to 3.9.

24. A process for preparing a composition as claimed in any one of claims 1 to 15 which comprises mixing a calcium compound with an acidulant so that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and the effective pH of the composition is from 3.5 to 4.5, obtaining an effective pH within the range 3.5 to 4.5 by addition of an alkali, if necessary or desired.

25 25. A method of reducing the tooth erosion properties of a solid or semi-solid acidic oral composition comprising adding calcium to the acidic oral composition so that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and the effective pH is from 3.5 to 4.5, obtaining an effective pH within the range 3.5 to 4.5 by addition of an alkali, if necessary or desired.

30 26. A method of reducing tooth erosion caused by acid in orally administered compositions comprising orally administering a solid or semi-solid composition comprising a calcium compound and an acidulant, characterised in that calcium is

present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective pH of the composition is from 3.5 to 4.5.

- 5 27. The use of a solid or semi-solid composition comprising a calcium compound and an acidulant in the manufacture of a medicament for the reduction of tooth erosion caused by acid in orally administered compositions, characterised in that calcium is present in the range of 0.3 to 0.8 mol per mol of acid and that the amount of calcium and acidulant in the composition is selected so that the effective
- 10 pH of the composition is from 3.5 to 4.5.

A. CLASSIFICATION F SUBJECT MATTER
IPC 6 A23L1/304 A61K7/16

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A23L A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P, X	WO 97 30601 A (PARKER DAVID MYATT ; SMITHKLINE BEECHAM PLC (GB)) 28 August 1997 cited in the application see page 5, line 18 - line 21; examples 1-3	1-27
X	US 5 028 446 A (SALEEB FOUAD Z ET AL) 2 July 1991 see column 3, line 36-49	1-15
X	EP 0 227 174 A (PROCTER & GAMBLE) 1 July 1987	1-15
Y	see page 2, line 33 - line 38; example 2	1-27
	-/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

15 January 1999

Date of mailing of the international search report

04.02.99

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3018

Authorized officer

Bendl, E

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 88 03762 A (UNIV TEXAS ;MISSION PHARMA CO (US)) 2 June 1988 cited in the application see table 2 -----	1-15
Y	GB 1 516 525 A (PROCTER & GAMBLE) 5 July 1978 see page 1, line 19 - line 25 see page 5, line 58 - line 62 -----	1-27
P,X	WO 98 13012 A (ENAMELON INC) 2 April 1998 see page 20, line 16 - line 19 see page 15, line 27 - page 16, line 26 -----	1-27
P,X	WO 98 13013 A (ENAMELON INC) 2 April 1998 see page 6, line 14 - line 20 see page 16, line 12 - line 13 -----	1-27

INTERNATIONAL SEARCH REPORT

International application No.
PCT/EP 98/05119

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☒ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:
see FURTHER INFORMATION sheet PCT/ISA/210
2. ☐ Claims Nos.:
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:
3. ☐ Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Although claims 16, 26, 27 and their dependent claims are directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound/composition.

Rule 39.1(iv) PCT - Method for treatment of the human or animal body by therapy

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 9730601	A	28-08-1997	AU 1872597 A	10-09-1997
			EP 0874558 A	04-11-1998
			FI 981781 A	19-08-1998
			GB 2326596 A	30-12-1998
			SE 9802763 A	03-09-1998
US 5028446	A	02-07-1991	CA 1309726 A	03-11-1992
			CN 1040023 A, B	28-02-1990
EP 0227174	A	01-07-1987	US 4737375 A	12-04-1988
			CA 1293641 A	31-12-1991
			DK 627986 A	27-06-1987
			GR 3001753 T	23-11-1992
			IE 61364 B	02-11-1994
			JP 1931531 C	12-05-1995
			JP 5036020 B	28-05-1993
			JP 63157964 A	30-06-1988
			PT 83983 B	17-01-1989
WO 8803762	A	02-06-1988	US 4851221 A	25-07-1989
			AU 605819 B	24-01-1991
			AU 8334787 A	16-06-1988
			CA 1297034 A	10-03-1992
			DE 3783610 A	25-02-1993
			DK 169231 B	19-09-1994
			EP 0329708 A	30-08-1989
			JP 2501619 T	07-06-1990
			JP 2614909 B	28-05-1997
GB 1516525	A	05-07-1978	AT 349641 B	10-04-1979
			AT 772275 A	15-09-1978
			AU 8534175 A	07-04-1977
			BE 834338 A	09-04-1976
			DE 2543489 A	22-04-1976
			FR 2287234 A	07-05-1976
			IE 42129 B	04-06-1980
			JP 51091339 A	10-08-1976
			NL 7511850 A	13-04-1976
			SE 7511327 A	12-04-1976
WO 9813012	A	02-04-1998	AU 4158297 A	17-04-1998
WO 9813013	A	02-04-1998	AU 4083597 A	17-04-1998